[a] θ_0^S refers to the Debye temperature at 0°K or to a temperature which is as close to 0°K as possible. If the temperature is different from 0°K, it is noted in a footnote.

[b] θ_{298}^S refers to the Debye temperature at 298°K as calculated from the specific heat at constant volume (Table XIV), unless otherwise noted. The values taken from the references cited are generally for Debye temperatures determined at $\sim\frac{1}{2}\theta^S$.

[c] θ^M refers to the Debye temperature calculated from the Lindemann equation. using the constant 138.5; see text for further discussion.

[d] Calculated by Kaufman and Clougherty[4] from the data of Johnston *et al.*[108] This value applies to a temperature of 13°K.

[e] See text for further discussion.

[f] Estimated value; see text for further discussion.

[g] This value applies to 40°K. From an extrapolation of high-temperature data to 0°K a value of 105°K is obtained for θ_0^S.

[h] Rajdev and Whitmore[107] re-evaluated Martin's data[14] and obtained a value of $\theta_0^S = 220$ which lies within the error listed above.

[i] Mean value of Debye temperature calculated from C_v (Table XIV) and Debye temperatures given by Murao[103] and Arajs and Colvin.[102]

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